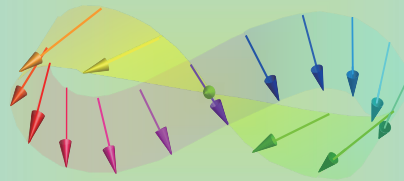


# Nanoelectronics

## 13



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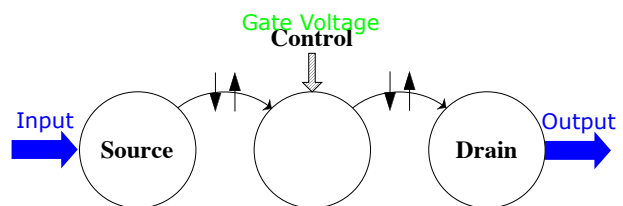
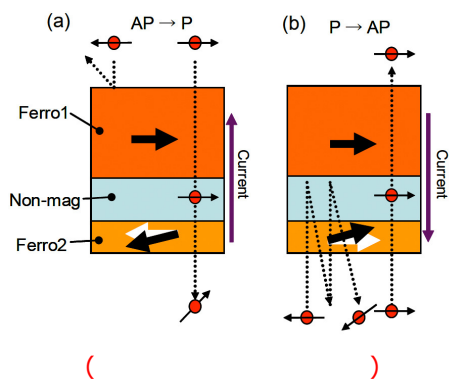
THE UNIVERSITY of York



09:00 Thursday, 02/March/2023 (P/T 005A)



### Quick Review over the Last Lecture



	FM / SC hybrid Structures	Magnetic tunnel junctions (MTJ)	All metal and spin valve structures
Interface			
Spin carriers			
Device applications	FM / 2DEG Schottky diodes Spin FET Spin LED Spin RTD	MOS junctions Coulomb blockade structures SP-STM Supercond. point contacts Spin RTD	Johnson transistors Spin valve transistors

\* After M. Johnson, *IEEE Spectrum* **37**, 33 (2000).



# Contents of Nanoelectronics

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- I. Introduction to Nanoelectronics (01)
  - 01 Micro- or nano-electronics ?
- II. Electromagnetism (02 & 03)
  - 02 Maxwell equations
  - 03 Scalar and vector potentials
- III. Basics of quantum mechanics (04 ~ 06)
  - 04 History of quantum mechanics 1
  - 05 History of quantum mechanics 2
  - 06 Schrödinger equation
- IV. Applications of quantum mechanics (07, 10, 11, 13 & 14)
  - 07 Quantum well
  - 10 Harmonic oscillator
  - 11 Magnetic spin
  - 13 Quantum statistics 1
- V. Nanodevices (08, 09, 12, 15 ~ 18)
  - 08 Tunnelling nanodevices
  - 09 Nanomeasurements
  - 12 Spintronic nanodevices

## 13 Quantum Statistics 1

- Classical distribution
- Fermi-Dirac distribution
- Bose-Einstein distribution



## Classical Distribution

Maxwell-Boltzmann distribution :

In the classical theory, the same types of particles can be distinguished.

→ Every particle independently fills each quantum state.

In order to distribute  $N$  particles to the states 1, 2, ... with numbers of  $N_1, N_2, \dots$

$$\frac{N!}{N_1!N_2!N_3!\dots}$$

To distribute  $N_i$  particles to the states  $G_i$  is  $G_i^{N_i}$ , and hence the whole states are

$$G_1^{N_1} G_2^{N_2} G_3^{N_3} \dots$$

Microscopic states in the whole system  $W$  are

$$W = \frac{N!}{N_1!N_2!N_3!\dots} \cdot G_1^{N_1} G_2^{N_2} G_3^{N_3} \dots$$

By taking logarithm for both sides and assuming  $N_i$  and  $G_i$  are large, we apply the Stirling formula.

$$\ln W = N \ln N - \sum_i N_i \ln \frac{N_i}{G_i} = \left( \sum_i N_i \right) \ln \left( \sum_i N_i \right) - \sum_i N_i \ln \frac{N_i}{G_i} \begin{cases} \ln G_i \cong G_i \ln G_i - G_i \\ \ln N_i! \cong N_i \ln N_i - N_i \end{cases}$$



## Classical Distribution (Cont'd)

In the equilibrium state,

$$\frac{\partial}{\partial N_i} \left( \ln W - \alpha \sum_i N_i - \beta \sum_i E_i N_i \right) = 0$$

Again, by applying the Stirling formula,  $\begin{cases} \ln G_i \cong G_i \ln G_i - G_i \\ \ln N_i! \cong N_i \ln N_i - N_i \end{cases}$

$$\ln \frac{N_i}{G_i} - \ln N + \alpha + \beta E_i = 0$$

Here, with using

$$\ln W = \sum_i \left[ \pm G_i \ln G_i \mp (G_i \mp N_i) \ln (G_i \mp N_i) - N_i \ln N_i \right]$$

$$\therefore \frac{N_i}{G_i} = N \exp(-\alpha - \beta E_i)$$

By using  $N = \sum_i N_i$  and  $\frac{N_i}{G_i} = \frac{1}{\exp(\alpha + \beta E_i) \pm 1}$

$$N = N \exp(-\alpha) \sum_i G_i \exp(-\beta E_i)$$

Now, the partition function is

$$Z = \sum_i G_i \exp(-\beta E_i)$$



# Classical Distribution (Cont'd)

Therefore,

$$\exp(-\alpha) = \frac{1}{Z}$$

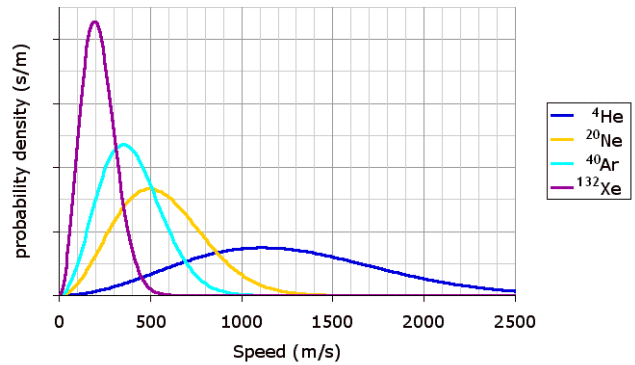
By substituting this relationship into  $\frac{N_i}{G_i} = N \exp(-\alpha - \beta E_i)$

$$\frac{N_i}{N} = \frac{G_i \exp(-\beta E_i)}{Z}$$

Here,  $\beta = 1/k_B T$

$G_i$  : degree of degeneracy of eigen energy  $E_i$

Maxwell-Boltzmann Molecular Speed Distribution for Noble Gases

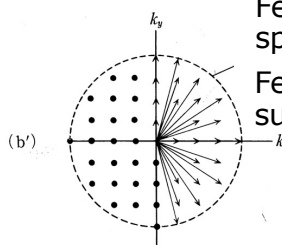
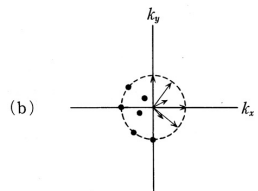
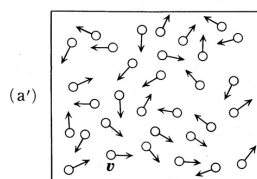
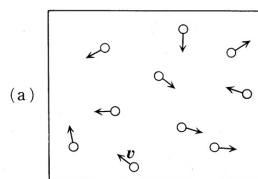


\* <http://www.wikipedia.org/>



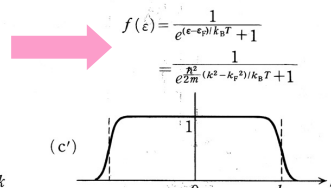
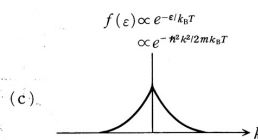
# Fermi-Dirac / Maxwell-Boltzmann Distribution

Electron number density :



Fermi sphere : sphere with the radius  $k_F$

Fermi surface : surface of the Fermi sphere



$$f(\epsilon) \propto e^{-\epsilon/k_B T}$$

$$\propto e^{-\hbar^2 k^2 / 2m k_B T}$$

$$f(\epsilon) = \frac{1}{e^{(\epsilon - \epsilon_F)/k_B T} + 1}$$

$$= \frac{1}{e^{\frac{\hbar^2 (k^2 - k_F^2)}{2m k_B T} + 1}}$$

distribution

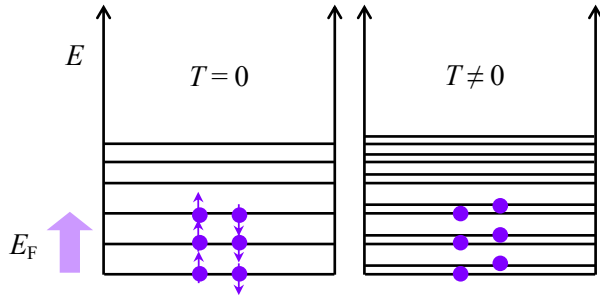
distribution

( electron number density ) ( electron number density )



# Fermi Energy

Fermi-Dirac distribution :



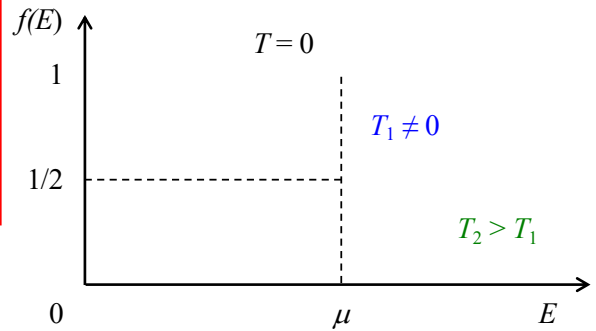
Pauli exclusion principle

At temperature  $T$ , probability that one energy state  $E$  is occupied by an electron :

$$f(E) =$$

$\mu$  : chemical potential  
 (= Fermi energy  $E_F$  at  $T=0$ )

$k_B$  : Boltzmann constant



# Fermi velocity and Mean Free Path

Fermi wave number  $k_F$  represents  $E_F$  :

$$\text{Fermi velocity : } v_F = \sqrt{\frac{2E_F}{m}}$$

$$k_F = \frac{m}{\hbar} v_F = \frac{\sqrt{2mE_F}}{\hbar}$$

Under an electrical field :

Electrons, which can travel, has an energy of  $\sim E_F$  with velocity of  $v_F$

For collision time  $\tau$ , average length of electron path without collision is

$$\ell = v_F \tau$$

Density of states :

Number of quantum states at a certain energy in a unit volume

$$g(E) = 2 \frac{1}{(2\pi)^3} \frac{4\pi}{2} \left( \frac{2m}{\hbar^2} \right)^{3/2} \sqrt{E} dE$$

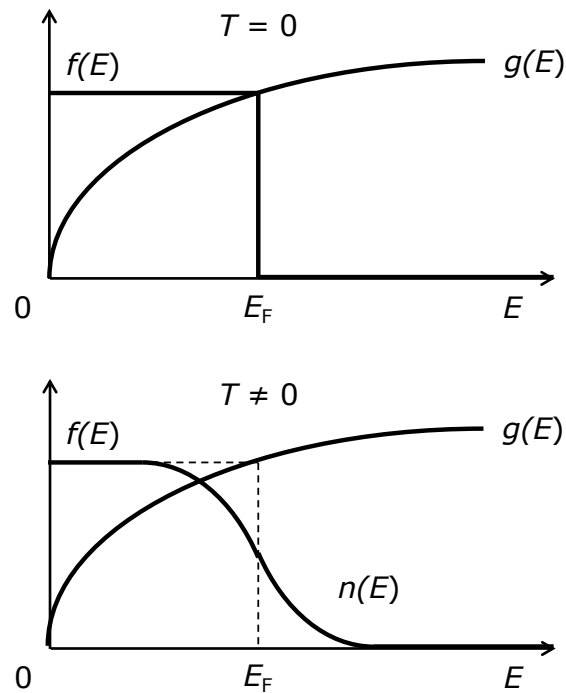




## Density of States (DOS) and Fermi Distribution

Carrier number density  $n$  is defined as :

$$n = \int f(E)g(E)dE$$



## Fermi-Dirac Distribution

In the Fermi-Dirac distribution, 1 quantum state can be filled by 1 particle.

→ Fermi particle (spin 1/2)

For the entire wavefunction :

$$\psi(\dots, x_i, \dots, x_j, \dots) = -\psi(\dots, x_j, \dots, x_i, \dots)$$

→ Particle exchange induces a sign change of the wavefunction.

For a 2-Fermi-particle system, due to the exchange characteristics, the wavefunction is expressed as

$$\psi(x_1, x_2) = \phi(x_1)\chi(x_2) - \phi(x_2)\chi(x_1)$$

If the 2 particles satisfy the same wavefunction,

$$\psi(x_1, x_2) = \phi(x_1)\chi(x_2) - \phi(x_1)\chi(x_2) = 0$$

→ principle



# Bose-Einstein Distribution

In the Bose-Einstein distribution, 1 quantum state can be filled by many particles.

→ Bose particle (spin 1,2,...)

For the entire wavefunction :

$$\psi(\dots, x_i, \dots, x_j, \dots) = \psi(\dots, x_j, \dots, x_i, \dots)$$

→ Particle exchange does not induce a sign change of the wavefunction.

For a 2-Bose-particle system, due to the exchange characteristics, the wavefunction is expressed as

$$\psi(x_1, x_2) = \phi(x_1)\chi(x_2) + \phi(x_2)\chi(x_1)$$

If the 2 particles satisfy the same wavefunction,

$$\psi(x_1, x_2) = \phi(x_1)\chi(x_2)$$

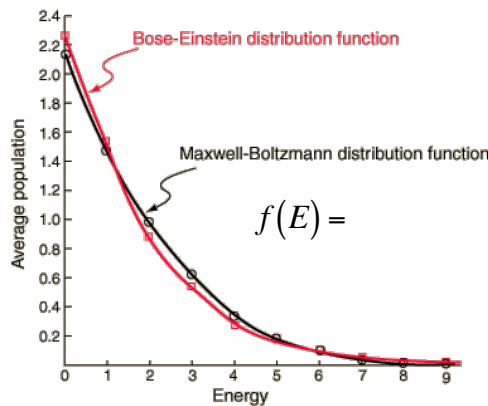
$$f(E) =$$



# Bose-Einstein Distribution (Cont'd)

Difference between Bose-Einstein and Maxwell-Boltzmann distributions :

$$f(E) =$$





# Quantum Statistics

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In quantum mechanics, the same types of particles cannot be distinguished :

By exchanging coordinate systems, Hamiltonian (operator) does not change :

$$HP = PH$$

$P$  : coordinate exchange operator

→ Diagonalization between  $H$  and  $P$

By assuming the eigenvalue for  $P$  to be  $\lambda$ ,

$$P\psi = \lambda\psi$$

Since twice exchange of the coordinate systems recover the original state,

$$P^2 = 1$$

$$\therefore P^2\psi = \psi = \lambda^2\psi$$

$$\therefore \lambda^2 = 1 \quad \therefore \lambda = \pm 1$$

Therefore,

$\psi = \psi$  (symmetric exchange) → Bose-Einstein statistics

→ spins

$\psi = -\psi$  (asymmetric exchange) → Fermi-Dirac statistics

→ spins