

Steps in the Exact Kohn-Sham Potential of Ensemble Density Functional Theory for Excited States and Their Relation to the Derivative Discontinuity

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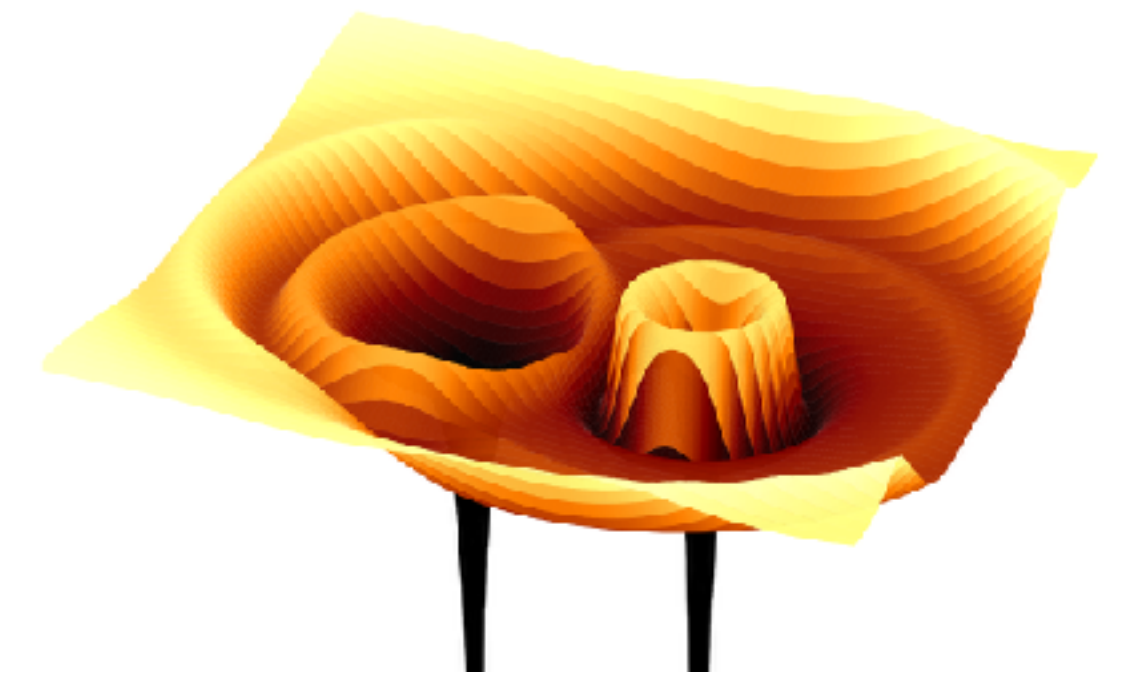
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LOS ANGELES MARCH 5-9



Ensemble Density Functional Theory

E. K. Gross, L. N. Oliveira, and W. Kohn, Phys. Rev. A **37**, 2809 (1988)

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The electron density of an *excited* system of N interacting electrons is modelled using an auxiliary system of non-interacting electrons:

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Electron density:
$$n(r) = \delta |\phi_{N+1}(r)|^2 + (1 - \delta) |\phi_N(r)|^2 + \sum_{i=1}^{N-1} |\phi_i(r)|^2$$

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The accuracy of the density relies on the approximation to the exchange-correlation part of v_s .

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The iDEA code

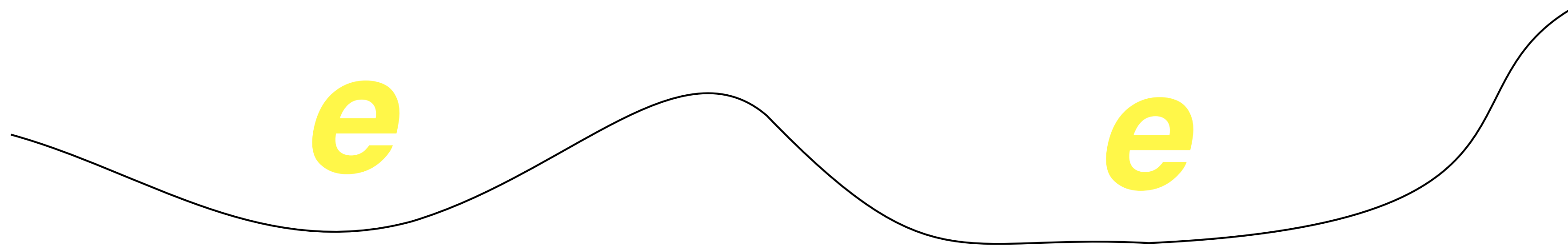


The iDEA code



Model simple 1D systems consisting of a few electrons

Choose any external potential we like

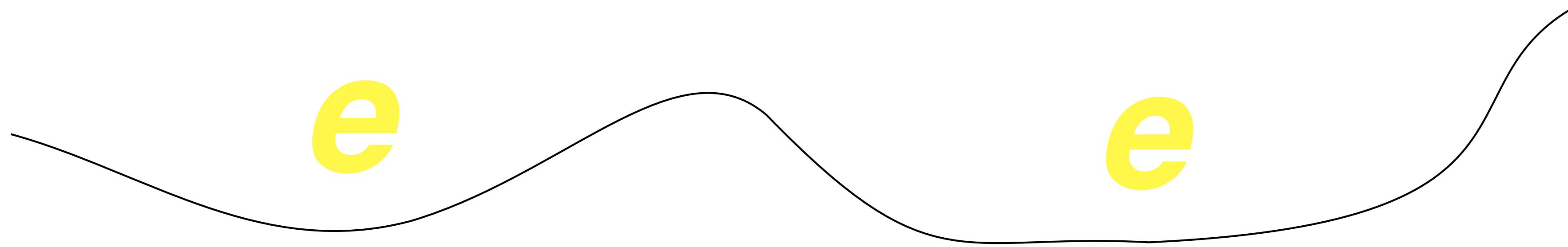


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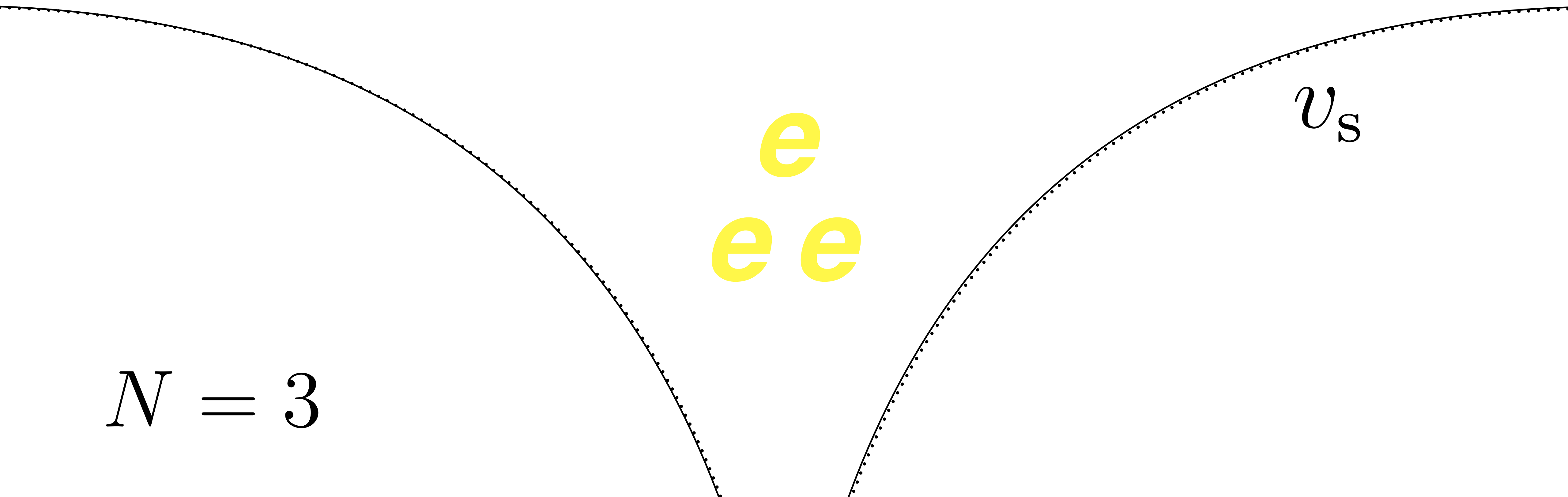
1. Calculate the exact ground-state and **first excited-state** electron density

2. The electron density is a linear combination of these two densities:

$$n(r) = (1 - \delta) \cdot n_0(r) + \delta \cdot n_1(r)$$

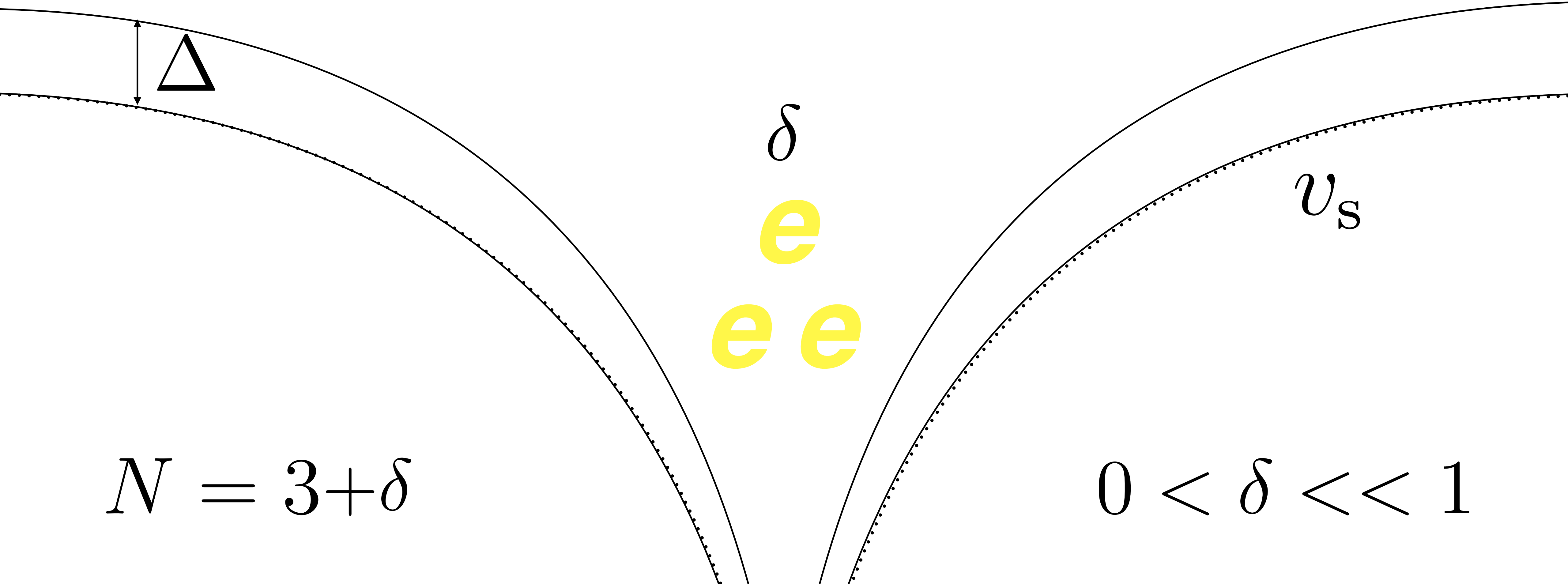
3. Next reverse-engineer the exact Kohn-Sham potential

Derivative discontinuity of an atom

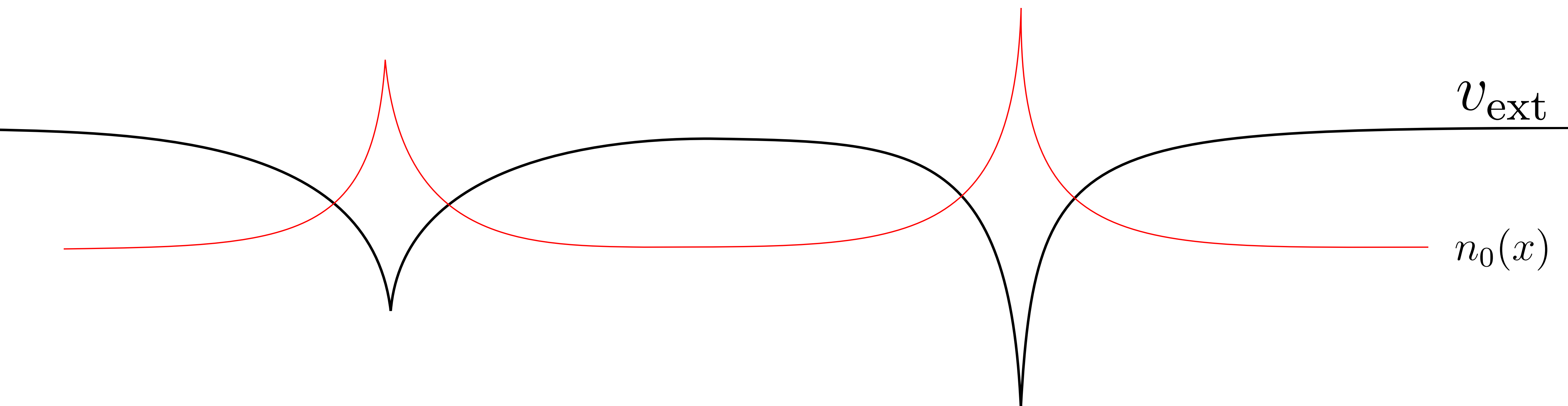


Derivative discontinuity of an atom

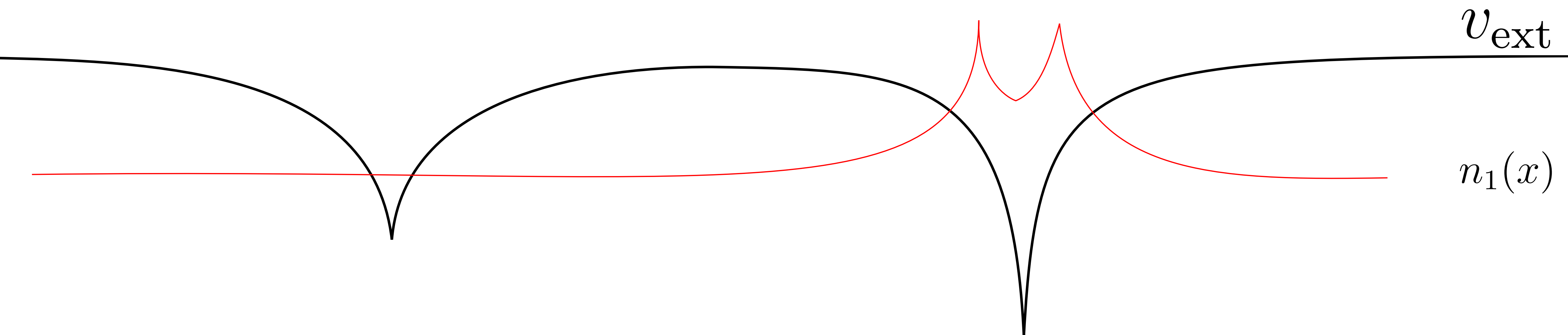
$$\Delta = I - A + \varepsilon^{\text{ho}} - \varepsilon^{\text{lu}}$$



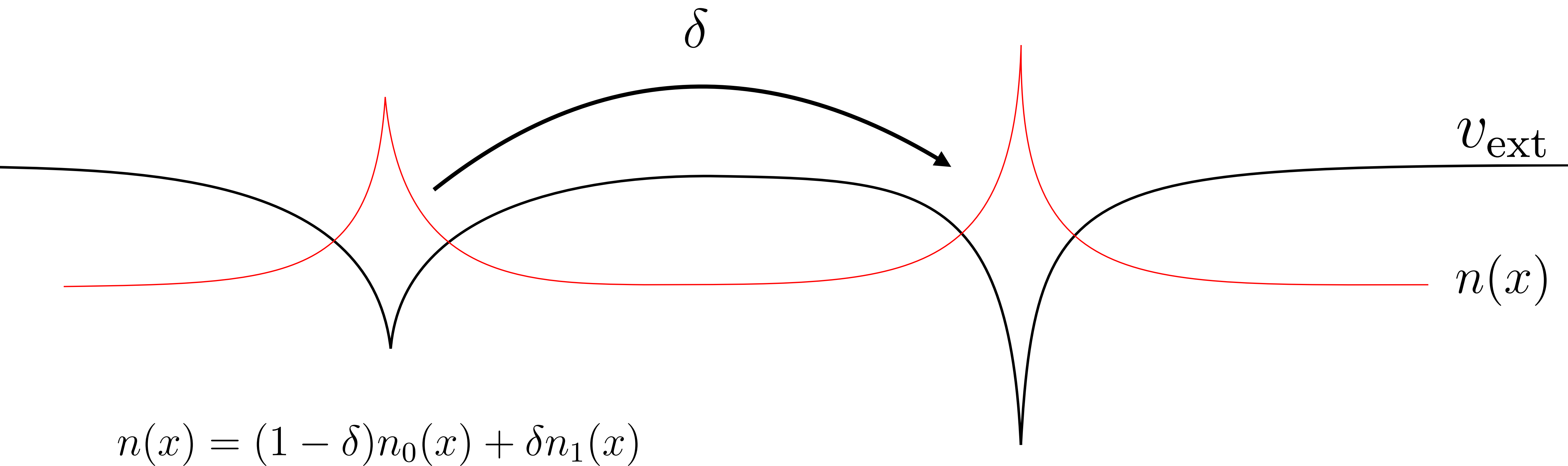
Charge transfer excitation in a 1D molecule



Charge transfer excitation in a 1D molecule



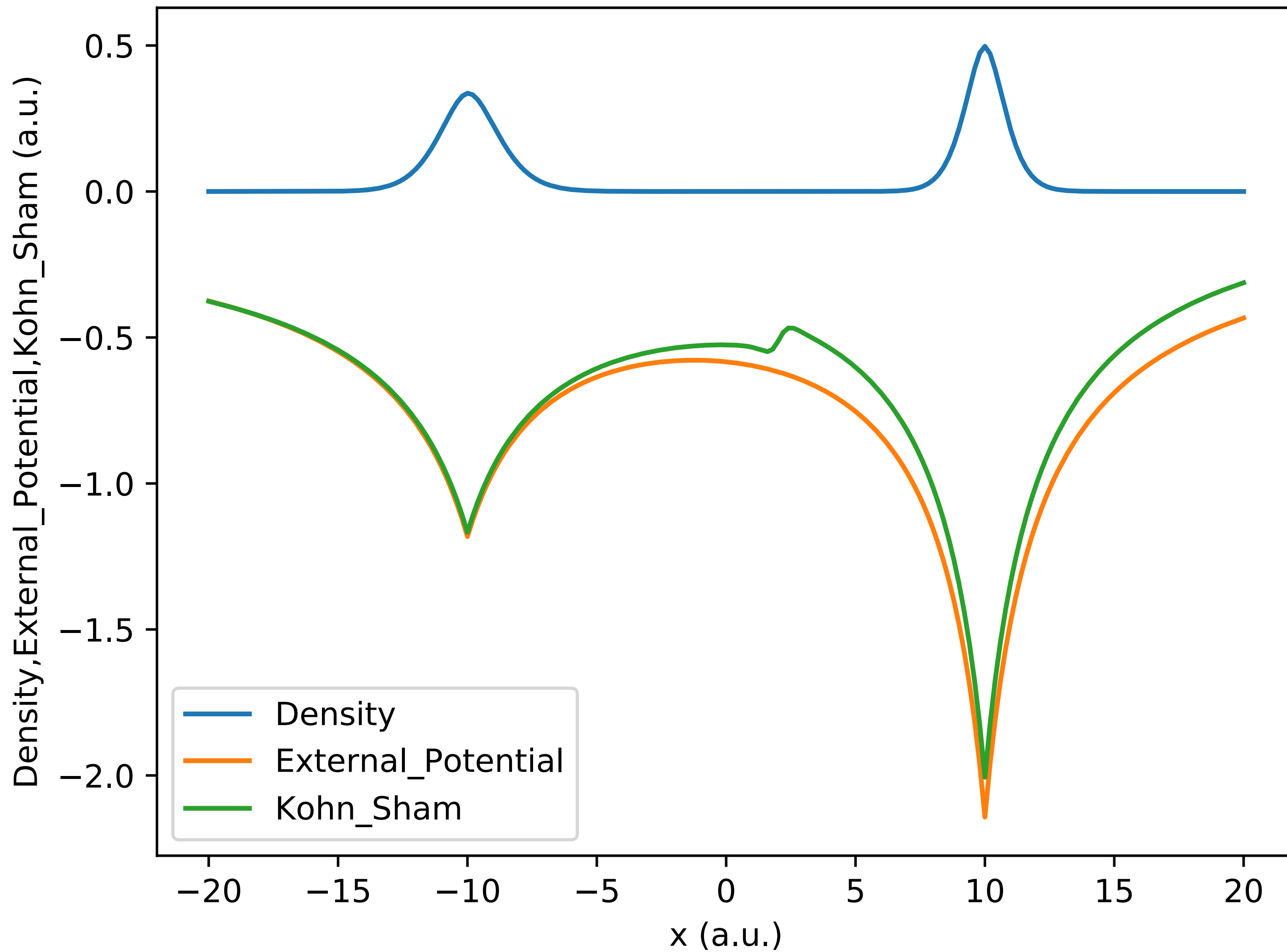
Charge transfer excitation in a 1D molecule



$$n(x) = (1 - \delta)n_0(x) + \delta n_1(x)$$

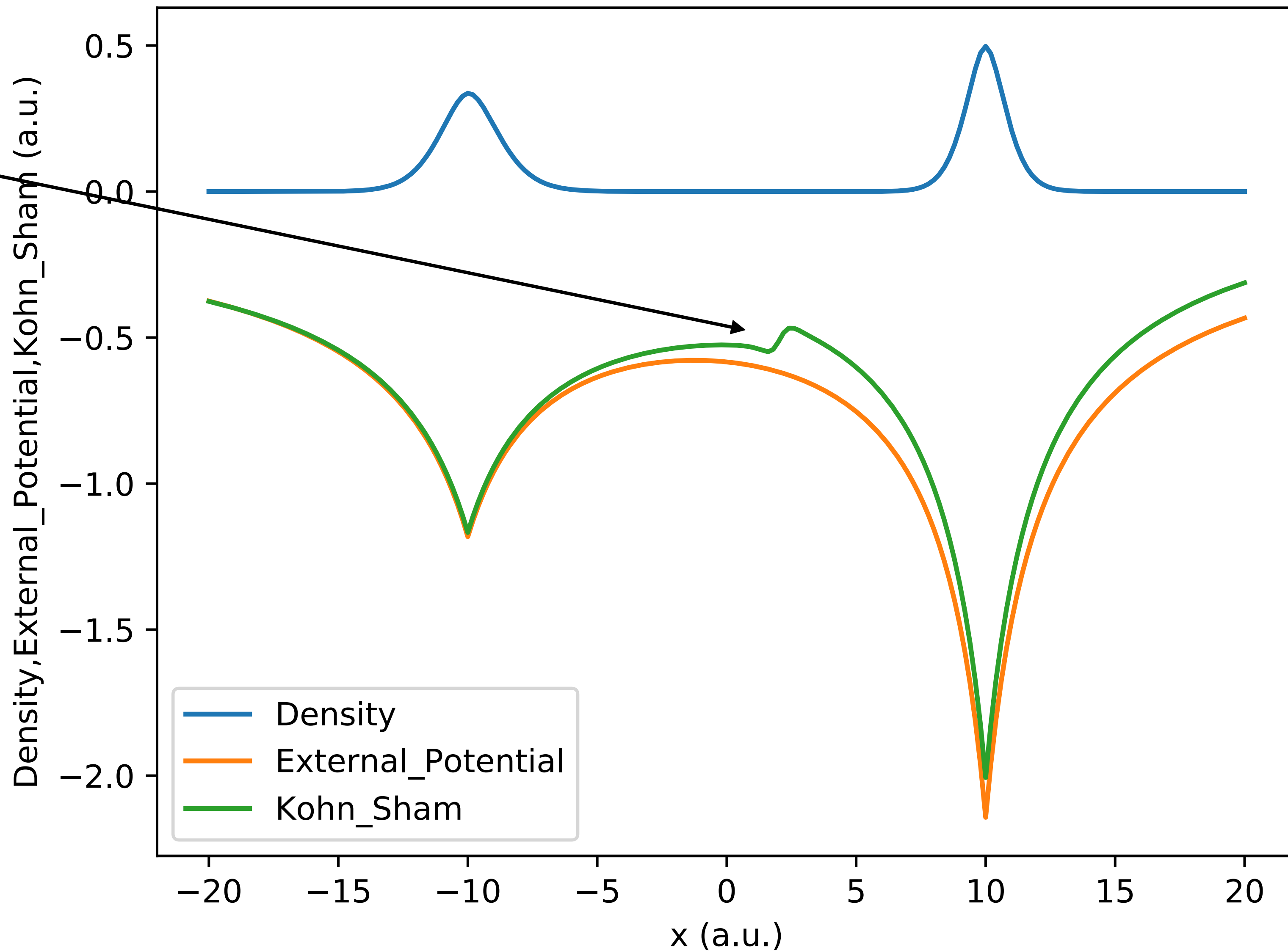
$$0 < \delta \ll 1$$

Ground-state 1D molecule



Ground-state 1D molecule

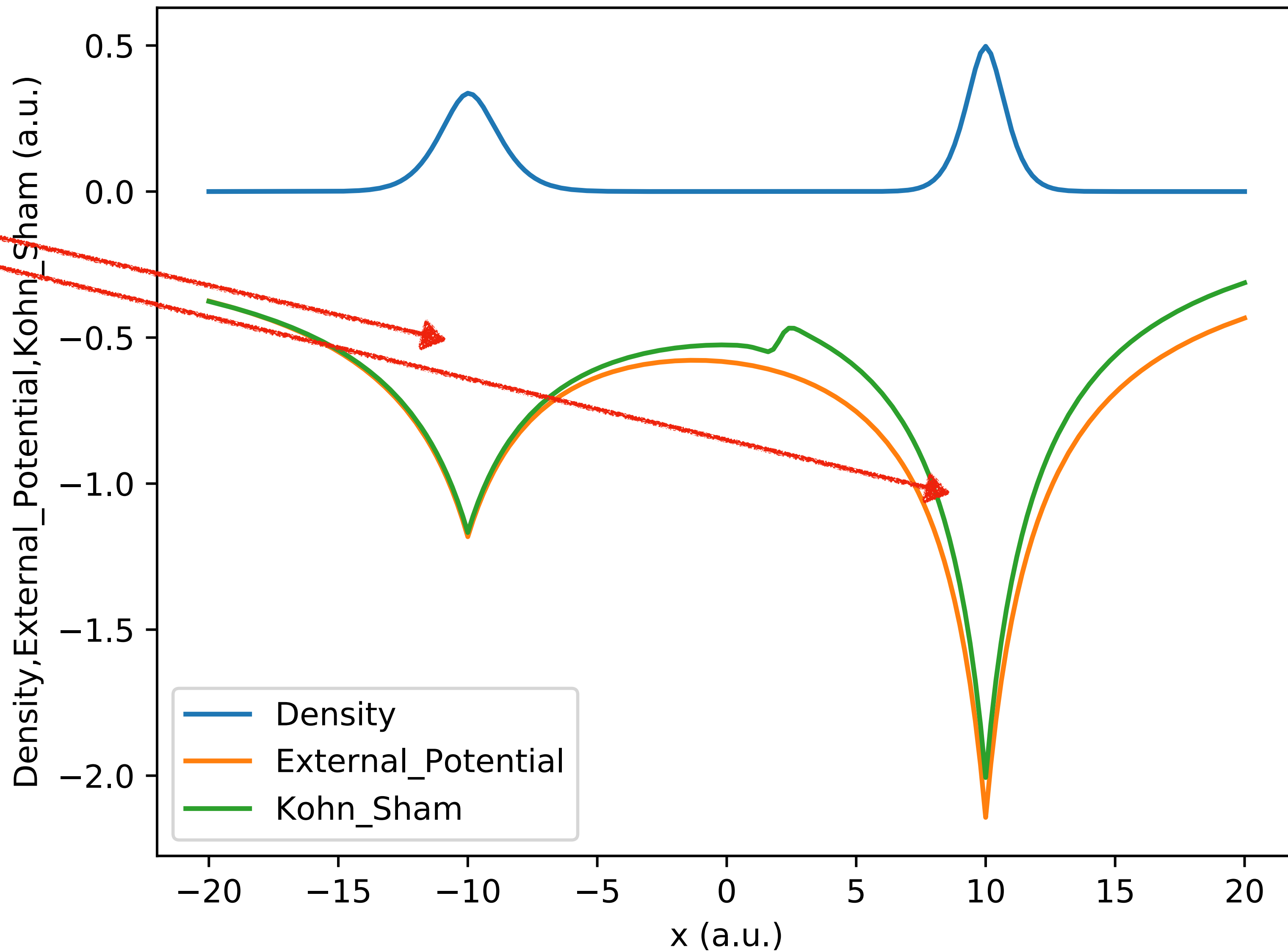
$$S = I_R - I_L + \eta_R^{\text{ho}} - \eta_L^{\text{ho}}$$



Ground-state 1D molecule

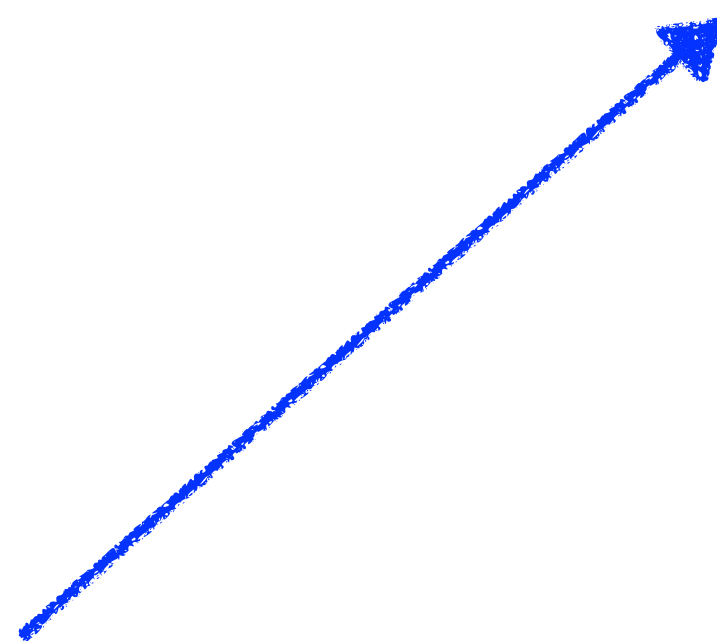
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Ionisation energy



Ground-state 1D molecule

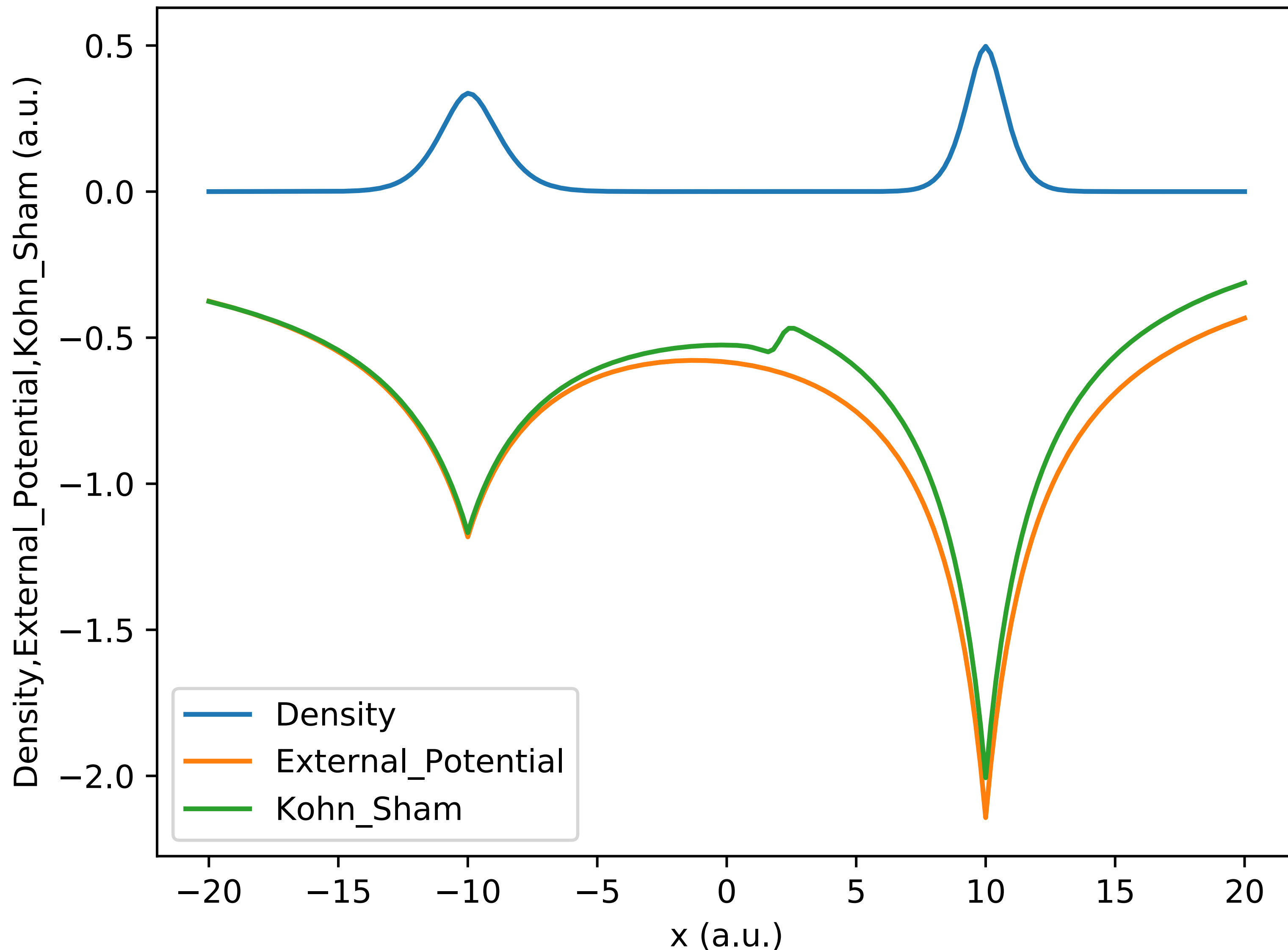
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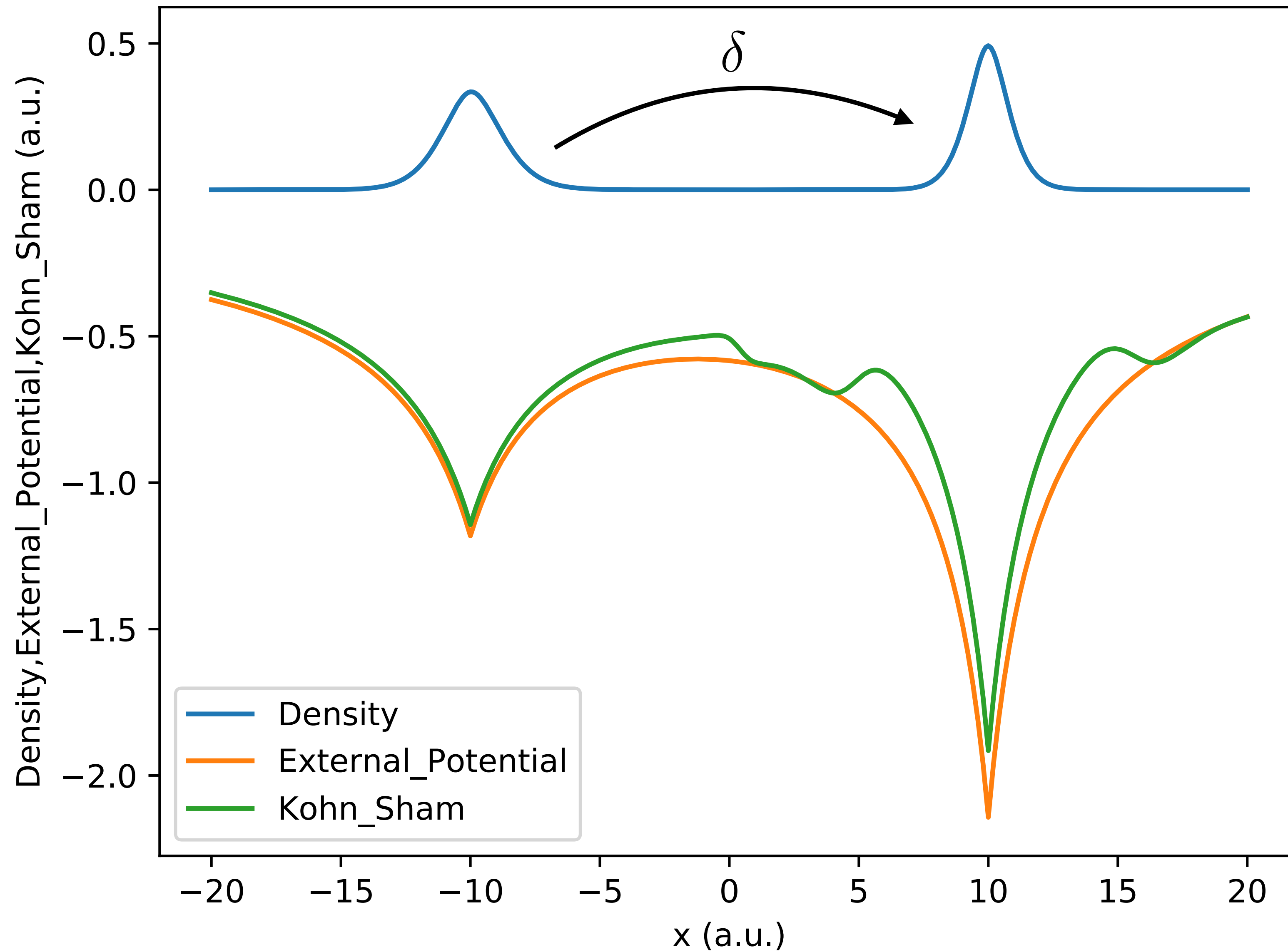
Kohn-Sham molecular energies

Bonded case: $\eta_R^{\text{ho}} = \eta_L^{\text{ho}}$

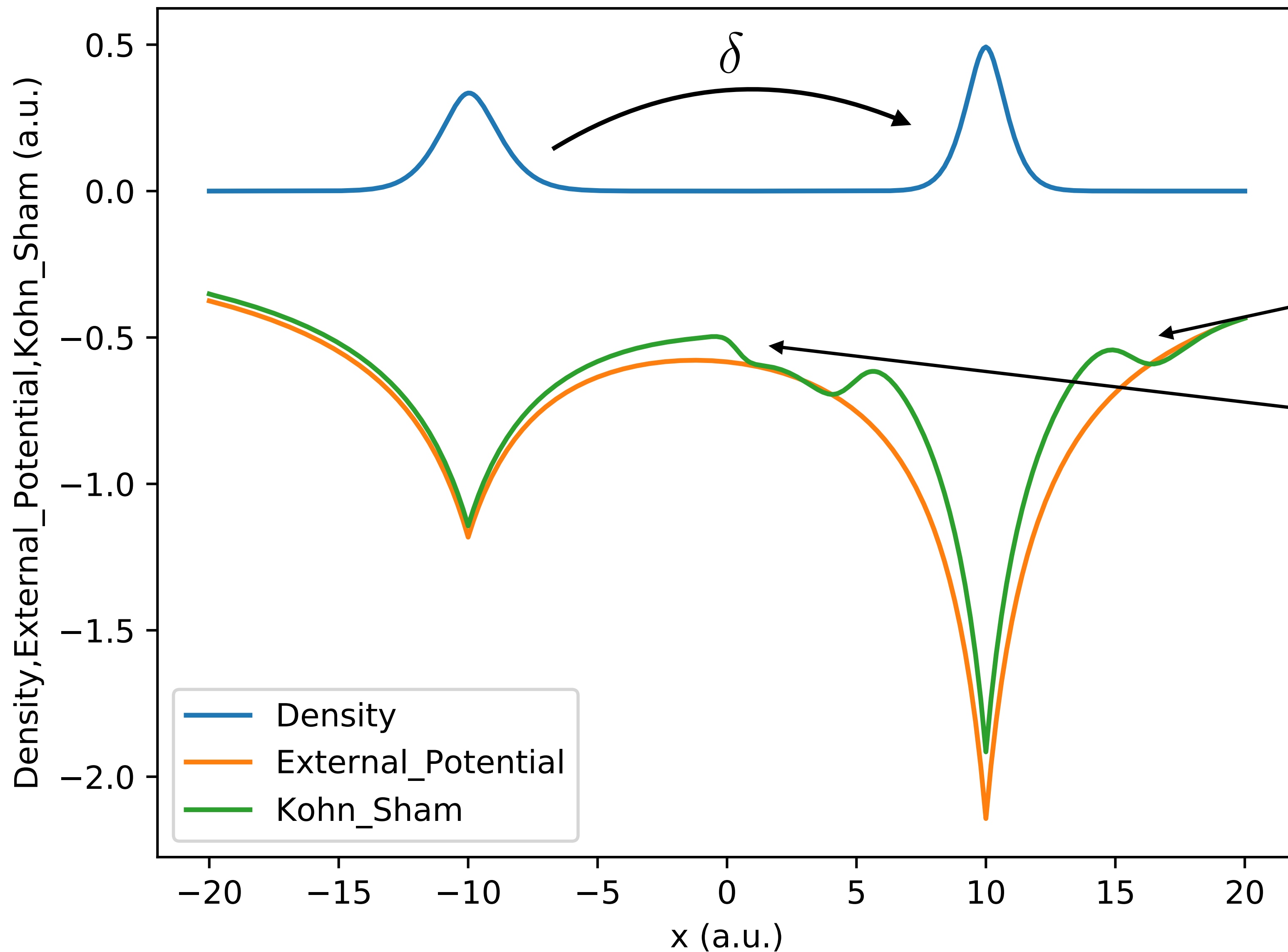
$$\Rightarrow S = I_R - I_L$$



Charge transfer excitation in a 1D molecule



Charge transfer excitation in a 1D molecule

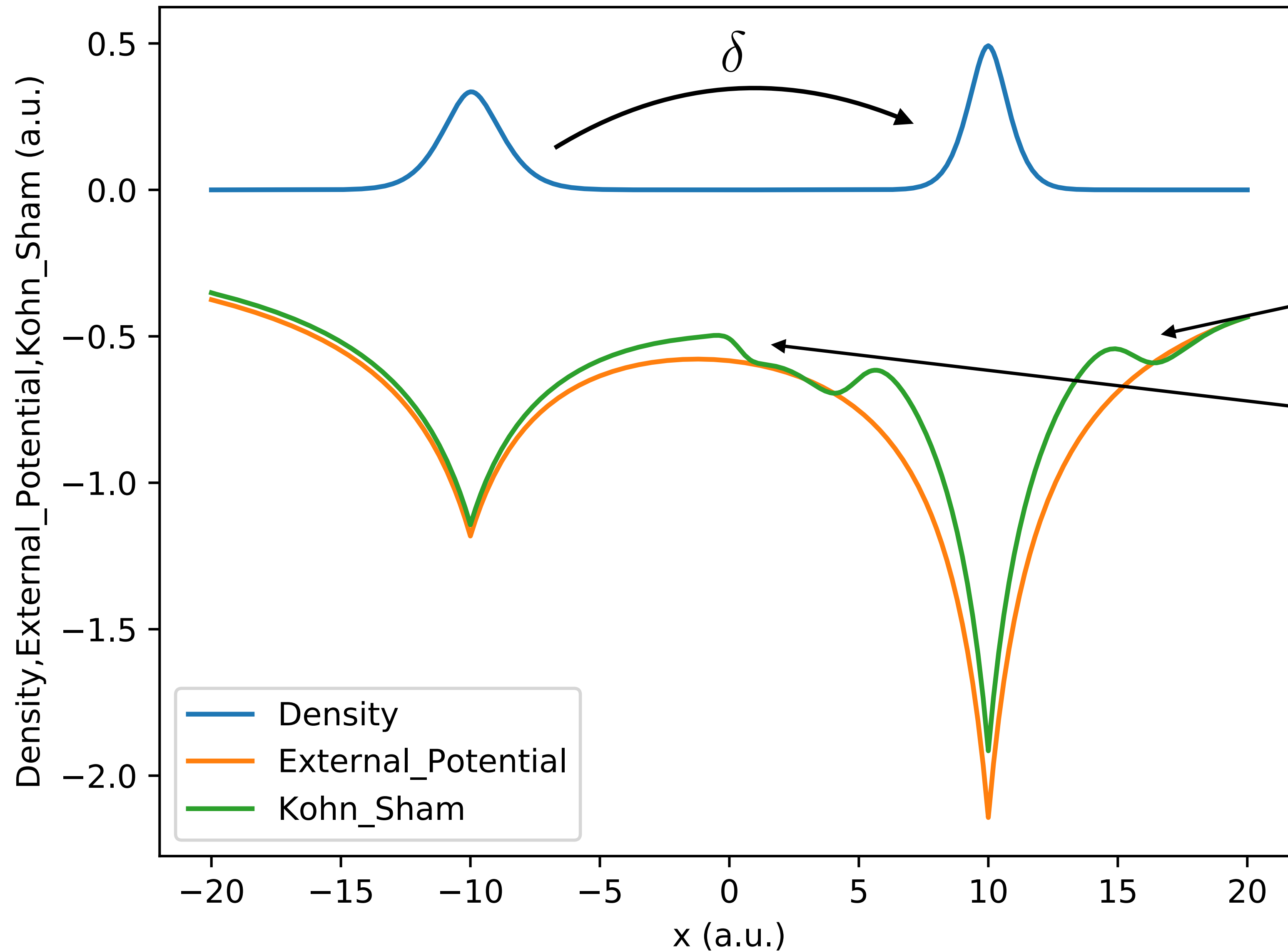


Derivative discontinuity of Atom R:

$$\Delta_R = I_R - A_R + \varepsilon_R^{\text{ho}} - \varepsilon_R^{\text{lu}}$$

$$\Delta_{L \rightarrow R}^{\text{CT}} = I_L - A_R + \eta_L^{\text{ho}} - \eta_R^{\text{lu}}$$

Charge transfer excitation in a 1D molecule



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What's this?

Charge-transfer derivative discontinuity

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Charge-transfer derivative discontinuity

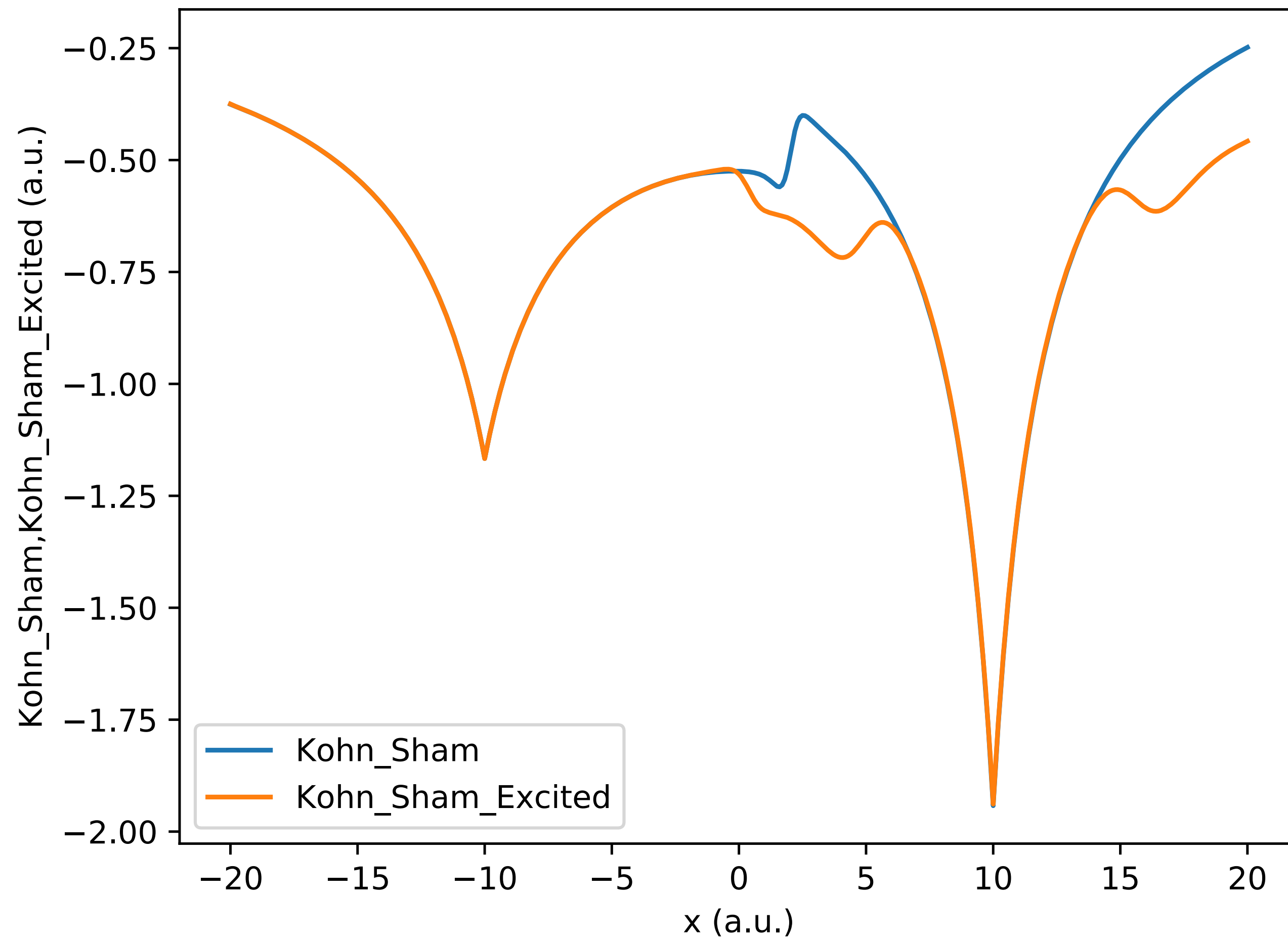
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Charge-transfer derivative discontinuity



$$S = S_1 + S_2 = \Delta_R - \Delta_{L \rightarrow R}^{\text{CT}} = I_R - I_L + \eta_R^{\text{ho}} - \eta_L^{\text{ho}}$$

Conclusions

1. Ensemble density functional theory can be used to model charge transfer
2. Upon charge transfer two plateaus form around the atoms of a diatomic molecule – one corresponds to the derivative discontinuity of the acceptor and the other corresponds to the ‘charge-transfer derivative discontinuity’
3. The steps which correspond to the two derivative discontinuities add together to make the interatomic step which determines the distribution of charge in the molecule

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Thanks for listening!

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