

Epitaxial growth and magnetic properties of half-metallic Fe₃O₄ on GaAs(100)

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The growth and magnetic properties of epitaxial magnetite Fe₃O₄ on GaAs(100) have been studied by reflection high-energy electron diffraction, x-ray photoelectron spectroscopy, magneto-optical Kerr effect, and x-ray magnetic circular dichroism. The epitaxial Fe₃O₄ films were synthesized by *in situ* post growth annealing of ultrathin epitaxial Fe films at 500 K in an oxygen partial pressure of 5×10^{-5} mbar. The XMCD measurements show characteristic contributions from different sites of the ferrimagnetic magnetite unit cell, namely, Fe_{id}³⁺, Fe_{oh}²⁺, and Fe_{oh}³⁺. The epitaxial relationship was found to be Fe₃O₄(100)(011)//GaAs(100)(010) with the unit cell of Fe₃O₄ rotated by 45° to match that of GaAs(100) substrate. The films show a uniaxial magnetic anisotropy in a thickness range of about 2.0–6.0 nm with the easy axes along the [0 $\bar{1}$ 1] direction of the GaAs(100) substrate.

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One of the major steps in developing next generation spintronic devices is the synthesis of magnetic/semiconductor hybrid materials with high efficient spin injection and the Curie temperature above room temperature.^{1–4} Spin injection and detection using ferromagnetic transition metals such as Fe, Co, and Ni have been demonstrated, but the efficiency is less than 1–2 % apart from the large effect observed in photoexcitation and photoemission experiments with circularly polarized light.^{5,6} On the other hand, based on easy integration with semiconductor substrate, the III-V based diluted magnetic semiconductors (DMS's) have been proposed as promising candidates but their Curie temperatures are well below room temperature so far.^{7–9} At the same time, half-metallic Fe₃O₄, called magnetite, has attracted great attention recently for spintronics as it has high polarization at the Fermi level¹⁰ and relatively high electronic conductivity at room temperature,¹¹ which is believed to benefit the injection of spin carriers into the semiconductors. Epitaxial growth of Fe₃O₄ on MgO,^{12–14} Al₂O₃,¹³ and Pt (Ref. 15) has been demonstrated by several groups and several interesting magnetic properties such as superparamagnetism,¹² slow saturation behavior,¹³ and local out-of-plane magnetic moments in zero field¹³ have been observed. However, in order to incorporate Fe₃O₄ into spintronic devices, synthesis of Fe₃O₄ on semiconductor substrates, such as GaAs, might be essential. In this Brief Report, we present the growth and the magnetic properties of epitaxial ultrathin Fe₃O₄ films on GaAs(100). As far as we know, this might be the first time that a half-metallic iron oxide/semiconductor hybrid structure has been synthesized.

There are basically two approaches of synthesis of Fe₃O₄ reported in the literature.^{10–16} The first approach is to deposit Fe on the substrate, then oxidize the epitaxial Fe with ox-

dizing agents such as O₂ and NO₂. The oxidation thickness of the Fe can be as thick as 56 Å for 1200 L (1 L = 1 Langmuir = 1×10^{-6} mbars) O₂ exposure.¹⁶ Another growth procedure is to epitaxially grow Fe₃O₄ samples directly by oxidizing the evaporated Fe atoms in an oxygen rich environment.¹³ We followed the first approach, growing epitaxial Fe films on GaAs substrates first and then oxidizing the films to form Fe₃O₄ in a UHV chamber. The undoped epitaxial GaAs(100) wafer was purchased from Wafer Technology, Ltd., with EJ Flat Option, i.e., the major flat is parallel to the [011] crystallographic direction. The substrate was chemically cleaned using an H₂SO₄:H₂O₂:H₂O (4:1:1) solution for 30 s, followed by deionized (DI) water rinsing and isopropyl alcohol (IPA) vapor cleaning. After being transferred into the UHV chamber, the substrate was annealed at 830 K for about 30 min until a sharp reflection high-energy electron diffraction (RHEED) pattern can be identified as shown in Fig. 1(a).

The Fe films were grown at a rate of approximately 2 Å/min, as monitored by a quartz microbalance, using an *e*-beam evaporator while the substrate temperature was held at room temperature. We would like to note that the nominal thicknesses of these predeposited Fe layers were quoted for the samples after oxidation. It has been shown that room-temperature growth could produce a ferromagnetic Fe film with a bulklike moment after the onset of the ferromagnetism, and the magnetic dead layer could be avoided by suppressing the formation of Fe-As nonmagnetic compounds at the interface.¹⁷ Figure 1(b) is the RHEED pattern of a 6.0 nm Fe film grown on the GaAs (100) substrate, which confirms the epitaxial growth of Fe on GaAs (100). From the figure, we can know that the Fe film takes a bcc structure with the epitaxial relationship of

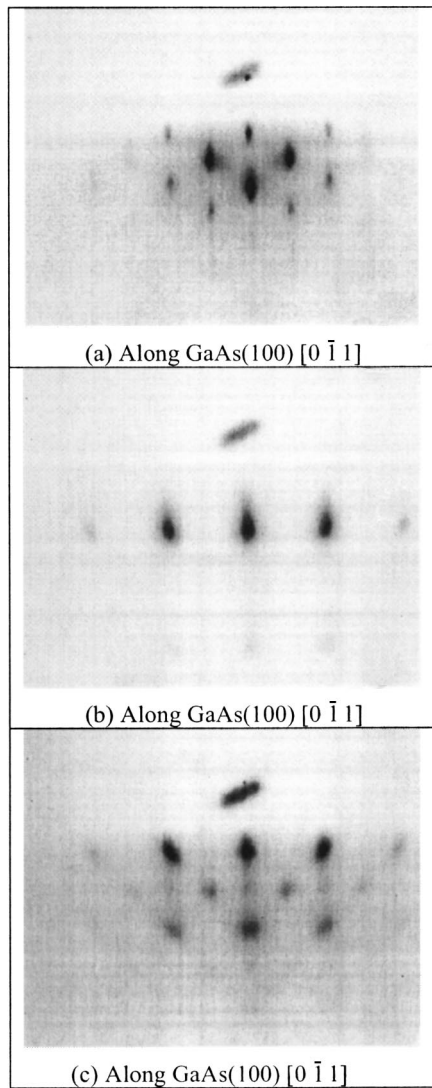


FIG. 1. RHEED patterns of (a) GaAs(100) substrate after cleaning and *in situ* annealing, (b) 6.0 nm of epitaxial Fe on GaAs(100), and (c) Fe₃O₄ formed by annealing at 500 K in an oxygen partial pressure. The electron beam was along the $[0\bar{1}1]$ direction of the GaAs(100) substrate and the beam energy was set at 10.0 kV in the above diffraction.

Fe(100) $\langle 001 \rangle$ //GaAs(100) $\langle 001 \rangle$. This alignment is consistent with other groups' reports and our previous work.¹⁷

After iron growth the sample was exposed to O₂ in the growth chamber, maintaining the O₂ partial pressure at 5×10^{-5} mbar with a leak valve. The distance from the O₂ nozzle to the sample was around 12 cm. This would allow formation of a uniform pressure around the sample. Figure 1(c) shows the RHEED pattern of the sample after 3 min of oxidization at a substrate temperature of 500 K. No further change of RHEED patterns was observed with further annealing at the same environment, suggesting that the sample had reached a stable structure. This RHEED pattern is the same as that observed in epitaxial Fe₃O₄/MgO(100) (Ref. 14) with the incident electron beam along the $[010]$ direction of Fe₃O₄, while in this work, the electron beam was along $[0\bar{1}1]$ direction of the GaAs(100) substrate. This

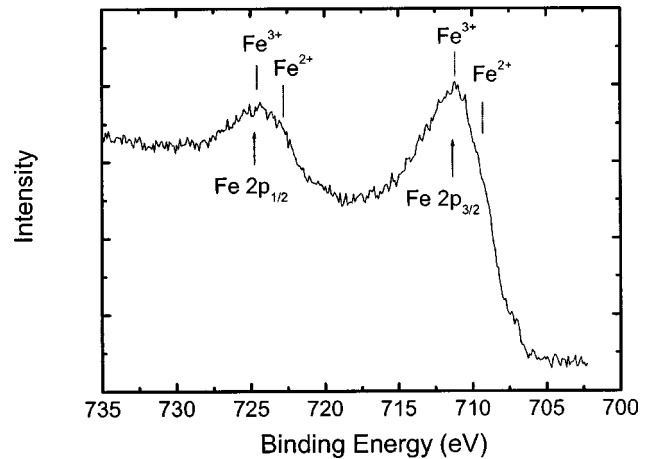


FIG. 2. Fe 2p core-level XPS spectroscopy from the 6.0 nm (nominal) sample excited by Mg- K_{α} . The x-ray is perpendicular to the surface of the sample.

suggests that in our work the $\langle 010 \rangle$ directions of the epitaxial magnetite cell parallel the $\langle 0\bar{1}1 \rangle$ directions of the GaAs(100) substrate, with an epitaxial relationship of Fe₃O₄(100) $\langle 011 \rangle$ //GaAs(100) $\langle 010 \rangle$.

As γ -Fe₂O₃ has a rather similar crystal structure to that of Fe₃O₄, the samples have been characterized using x-ray photoelectron spectroscopy (XPS). It is well established that the satellites in the XPS spectroscopy can help to identify the chemical states, for example, different chemical states of iron in its oxides.^{18–20} One remarkable difference between the γ -Fe₂O₃ (maghemite) and the Fe₃O₄ (magnetite) is that the former has satellites in the Fe 2p core level spectra while the latter does not have. Figure 2 shows the Fe 2p core-level spectroscopy of the 6.0 nm sample, obtained with normal emission using Mg- K_{α} radiation, which agrees well with the Fe₃O₄ spectra reported.^{18–20} The broad Fe 2p peaks are attributed to the coexistence of Fe³⁺ and Fe²⁺ states, and at the same time, that no satellite can be identified around the binding energy of 719.0 eV, excludes the possible presence of maghemite (γ -Fe₂O₃) in our samples. It is worthwhile to note that the XPS spectra along with the x-ray magnetic circular dichroism (XMCD) data in Fig. 5 show no sign of contribution from metallic Fe, indicating that these ultrathin Fe films were fully oxidized.

Figure 3 is a schematic diagram of the proposed alignment of the magnetite cell rotated by 45° relative to the GaAs (100) substrate. The rotation of 45° might be due to the fact that the Fe₃O₄(100) $\langle 011 \rangle$ directions, rather than Fe₃O₄(100) $\langle 010 \rangle$ directions, matches better with GaAs(100) $\langle 010 \rangle$ directions. The lattice constant of GaAs is 5.654 Å, and for Fe₃O₄, it is 8.396 Å. Simple calculation shows the mismatch between Fe₃O₄ and GaAs is around 5.0% if the Fe₃O₄ cell is rotated by 45° relative to GaAs cell in the (100) plane. This might be the reason that epitaxial magnetite can be stabilized on GaAs(100) as demonstrated here. However, we should note that although the relative rotation of the magnetite cells can be confirmed, it is not clear whether the Fe₃O₄ is rotated clockwise or anticlockwise relative to GaAs. The coherent Fe might be oxidized into one dominant rotated state due to the inverse spinel structure of the Fe₃O₄.

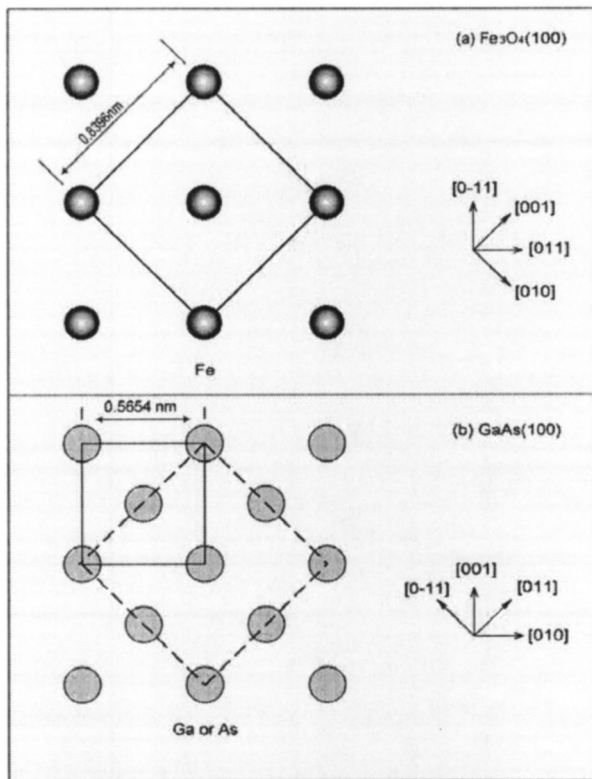


FIG. 3. Proposed epitaxial relation of (a) epitaxial Fe_3O_4 on (b) $\text{GaAs}(100)$. The unit cells and the lattice distances have been indicated by solid lines. The Fe_3O_4 unit cell has been rotated by 45° in the plane of (100) relative to GaAs . The dash line in (b) indicates one possible unit accommodating one Fe_3O_4 unit cell. Note in (a) only Fe ions in the bottom layer of a unit cell have been shown. Insets show major axes of both materials.

The magnetic properties of the films were studied by *ex situ* magneto-optical kerr effect (MOKE) at room temperature using an electromagnet with a maximum field of 1.6 kOe and a stabilized diode laser with a 780 nm wavelength. The samples were transferred in the air without a capping layer since Fe_3O_4 is one of the most stable iron oxides and any capping layer might change the magnetic properties of the films. The configuration of the MOKE was transverse in which the applied magnetic field was in the plane of the films. No MOKE signal was detected with thickness less than 1.0 nm, suggesting the films may be nonmagnetic at room temperature at the initial stage. At a narrow thickness range of around 1.6 nm, S-shaped MOKE loops with zero hysteresis were observed suggesting that the films are superparamagnetic as that found in Fe/GaAs system.¹⁷ The superparamagnetic behavior could be due to the formation of disconnected clusters, which are unable to stabilize a long-range ferromagnetic order at room temperature. The antiphase boundaries (APB's) among the clusters could also be responsible for this superparamagnetic behavior as suggested by Voogt *et al.*¹²

Magnetic uniaxial anisotropy was observed in a thickness range of about 2.0–6.0 nm as shown by Fig. 4 of the MOKE loops along their four major axes. At a thickness of 2.0 nm, a uniaxial magnetic anisotropy is clearly defined and around 4.2 nm, the system still exhibits a uniaxial anisotropy. While

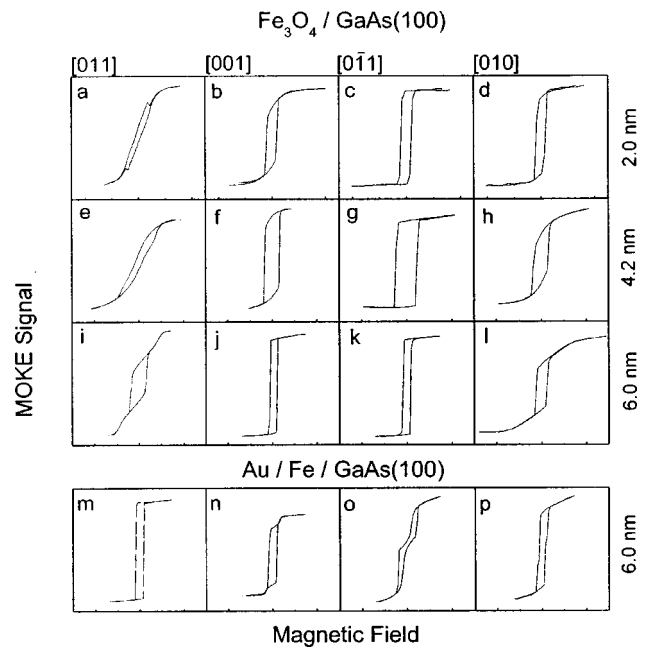


FIG. 4. MOKE loops of $\text{Fe}_3\text{O}_4/\text{GaAs}(100)$ [up panel, (a)–(l)] and $\text{Au}/\text{Fe}/\text{GaAs}(100)$ [bottom panel, (m)–(p)]. The nominal thicknesses of the samples are indicated to the right of the panels. The directions of the external magnetic field are labeled on top of each column according to the $\text{GaAs}(100)$ substrate. Note the magnetic field in the first row (a)–(d) was -0.25 – $+0.25$ kOe, while in other rows (e)–(p), it was -0.75 – $+0.75$ kOe.

at the thickness of 6.0 nm, the cubic anisotropy of the bulk Fe_3O_4 presents and rotates the global easy axis slowly away from the $[0\bar{1}1]$ direction. The easy axis of this uniaxial magnetic anisotropy (UMA) is along the $[0\bar{1}1]$ direction of the $\text{GaAs}(100)$ substrate, rotated by 90° compared with that in $\text{Au}/\text{Fe}(6.0 \text{ nm})/\text{GaAs}$ from the same growth.²¹ The origin of this uniaxial magnetic anisotropy might be related to the interface as that in the Fe/GaAs system. However, we would like to note that the origin of such uniaxial magnetic anisotropy in the $\text{Fe}/\text{GaAs}(100)$ remains an open issue. Two distinctly different mechanisms associated with “unidirectional chemical bonding” and “magnetoelastic coupling,” respectively, have been proposed to explain the uniaxial magnetic anisotropy in Fe/GaAs .^{22,23} These two mechanisms might be responsible for the uniaxial magnetic anisotropy observed here. The anisotropy field as estimated from the saturation fields along the hard axes is around 300 Oe, greatly reduced compared with the high saturation field of about 1.6 kOe observed in Ref. 13. This might be due to a possible decrease of anti-phase boundaries in the epitaxial $\text{Fe}_3\text{O}_4/\text{GaAs}$ structure.

The magnetic properties and the structure of the samples were further characterized using x-ray absorption spectroscopy (XAS) and XMCD, which are capable of probing the magnetic signal from sites of different chemical environments. Magnetite features an inverse spinel structure in which the tetrahedral sites are entirely occupied by $\text{Fe}_{\text{td}}^{3+}$ cations but the octahedral sites are equally filled up by $\text{Fe}_{\text{oh}}^{2+}$ and $\text{Fe}_{\text{oh}}^{3+}$ cations. Although the same amount of Fe^{3+} cations in tetrahedral sites and octahedral sites are ferrimagneti-

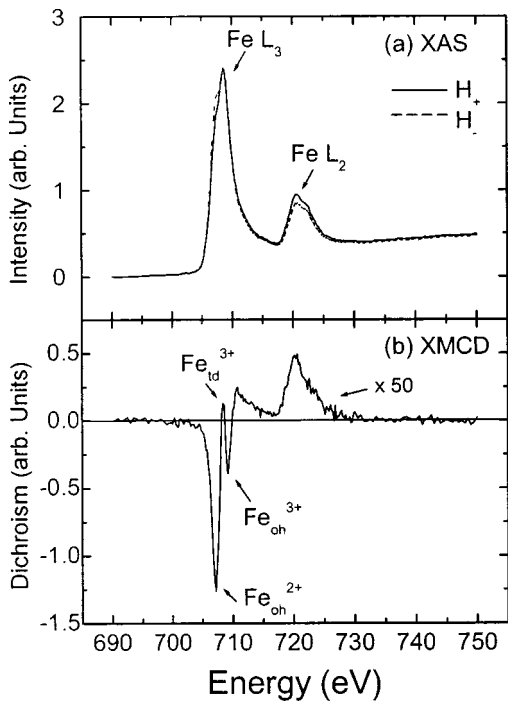


FIG. 5. (a) XAS and (b) XMCD of Fe_3O_4 (6.0 nm)/GaAs(100). Three peaks in the Fe L_3 edge related to the $\text{Fe}_{\text{td}}^{3+}$, $\text{Fe}_{\text{oh}}^{2+}$, and $\text{Fe}_{\text{oh}}^{3+}$ with different energy can be identified.

cally coupled, they can not cancel each other out in the dichroism due to their different chemical environments.²⁴ The measurements were carried out in station 1.1 in Daresbury in TEY mode. The 0.6 T magnetic field was parallel or antiparallel to the helicity of the incident x-ray, which was perpen-

dicular or 45° with the surfaces of the measured films. Figure 5 shows the XAS and XMCD spectra from a 6.0 nm Fe_3O_4 sample with the x-ray perpendicular to the sample. For the 45° setup, the spectra were similar. The characteristic contributions from different sites of $\text{Fe}_{\text{td}}^{3+}$, $\text{Fe}_{\text{oh}}^{2+}$, and $\text{Fe}_{\text{oh}}^{3+}$ in Fe_3O_4 can be clearly identified.²⁵ The strong peak of $\text{Fe}_{\text{oh}}^{2+}$ also excludes the composition of $\gamma\text{-Fe}_2\text{O}_3$, which does not contain any $\text{Fe}_{\text{oh}}^{2+}$. The macroscopic magnetization comes from the octahedral $\text{Fe}_{\text{oh}}^{2+}$ cations, which have different photon energies in the XMCD spectrum compared with the octahedral $\text{Fe}_{\text{oh}}^{3+}$ cations or tetrahedral $\text{Fe}_{\text{td}}^{3+}$ cations.

In summary, we have demonstrated the epitaxial growth of single-crystal Fe_3O_4 ultrathin films on GaAs(100) substrate by post growth annealing of ultrathin epitaxial Fe films in an oxygen partial pressure. The formation of the ferrimagnetic Fe_3O_4 rather than $\gamma\text{-Fe}_2\text{O}_3$ or other ion oxides has been confirmed by RHEED, XPS, and XMCD. The Fe_3O_4 /GaAs(100) system shows an interesting epitaxial relationship with the unit cell of the Fe_3O_4 rotated by 45° to match that of GaAs(100) substrate. The Fe_3O_4 films show a uniaxial magnetic anisotropy, unexpected from the crystal symmetry, with the easy axis along the $[0\bar{1}1]$ direction of the GaAs(100) substrate. While the understanding of the fundamental structure and magnetism of this magnetic/semiconductor system remains challenging and will attract further experimental and theoretical studies, Fe_3O_4 /GaAs(100) might be one of the most promising spintronic materials.

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